

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

### 3-(p-METHOXYBENZOYL)PROPIONIC ACID. AN IMPROVED SUCCINOYLATION PROCEDURE

Y. Shyamsunder Rao<sup>a</sup>; Richard A. Kretchmer<sup>a</sup>

<sup>a</sup> Department of Chemistry, Illinois Institute of Technology, Chicago, Illinois

**To cite this Article** Rao, Y. Shyamsunder and Kretchmer, Richard A.(1971) '3-(p-METHOXYBENZOYL)PROPIONIC ACID. AN IMPROVED SUCCINOYLATION PROCEDURE', *Organic Preparations and Procedures International*, 3: 4, 177 – 181

**To link to this Article:** DOI: 10.1080/00304947109356064

**URL:** <http://dx.doi.org/10.1080/00304947109356064>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

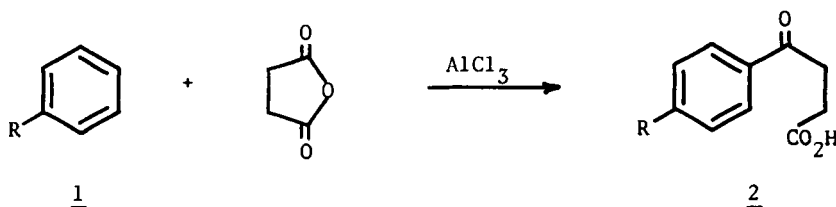
This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

3-(p-METHOXYBENZOYL)PROPIONIC ACID. AN IMPROVED SUCCINOYLATION PROCEDURE.

Y. Shyamsunder Rao and Richard A. Kretchmer<sup>1</sup>  
 Department of Chemistry  
 Illinois Institute of Technology  
 Chicago, Illinois 60616

The preparation of 3-(p-methoxybenzoyl)propionic acid (2, R = OCH<sub>3</sub>) by the Friedel-Crafts reaction between succinic anhydride and anisole in the presence of aluminum chloride is well documented in the literature.<sup>2</sup> This



reaction has been conducted in nitrobenzene,<sup>3</sup> nitropropane,<sup>4</sup> tetrachloroethane<sup>5</sup> and a tetrachloroethane-nitrobenzene mixture<sup>6</sup> to give 75-95% yields of product. Anisole,<sup>7</sup> carbon disulfide<sup>8</sup> and benzene<sup>9</sup> have also been used as solvents, but result in lower yields. In most of the reports, a lengthy reaction period (24-72 hr) was followed by hydrolysis and removal of solvent by steam distillation. This procedure is inconvenient, time consuming and solvent removal by steam distillation results in some degradation of the product.

We have found that long reaction times at low temperature are unnecessary and that the product can be isolated rapidly without the necessity of a steam distillation or extractive work-up. The time expended in completing the entire preparation is less than 3 hr. The results of a systematic study of

the succinoylation of anisole in various solvent systems using this procedure are indicated in Table I.

This succinoylation procedure was also found to be very satisfactory for several other monosubstituted benzenes (Table II). The success of this method is a reflection of the relative insolubility of the resulting  $\gamma$ -keto acids.

TABLE I  
Effect of Solvent in the Preparation of 3-(p-Methoxybenzoyl)-propionic Acid.

<u>Solvent</u>	<u>Isolated Yield, %</u>
Nitromethane	80
Nitroethane	72
Nitropropane	68
Tetrachloroethane-Nitromethane <sup>a</sup>	80
Tetrachloroethane-Nitroethane <sup>a</sup>	72
Tetrachloroethane-Nitropropane <sup>a</sup>	93
Tetrachloroethane-Nitrobenzene <sup>a</sup>	82

(a) 4:1 mixture by volume.

TABLE II  
 Succinoylation of Monosubstituted Benzenes in a 4:1<sup>a</sup> Tetrachloroethane-Nitrobenzene Solvent System

Reactant	Product	mp, °C <sup>b</sup>	lit. mp, °C	Isolated Yield, %
Anisole	3-(p-Methoxybenzoyl)propionic Acid	146	146.5-147 <sup>c</sup>	82
Toluene	3-(p-Methylbenzoyl)propionic Acid	124	127.5 <sup>d</sup>	84
Chlorobenzene	3-(p-Chlorobenzoyl)propionic Acid	131	131 <sup>e</sup>	37
Diphenyl ether <sup>f</sup>	3,3'-[Oxybis(p-phenylene-carbonyl)]-dipropionic Acid	221	217 <sup>g</sup>	94

(a) By volume. (b) Melting points are uncorrected. (c) Reference 6b. (d) Reference 10.

(e) Reference 11. (f) Two moles of succinic anhydride and five moles of anhydrous AlCl<sub>3</sub> were used per mole of diphenyl ether. (g) Reference 12.

### EXPERIMENTAL

Succinoylation of Anisole in Tetrachloroethane-Nitrobenzene. The following experimental is representative of the general procedure. A mixture of 10.8 g (0.10 mol) of anisole and 10.0 g (0.10 mol) of succinic anhydride was taken up in 80 ml of tetrachloroethane and 20 ml of nitrobenzene. To this was added, in small portions with stirring, 40.0 g (0.30 mol) of anhydrous aluminum chloride while the temperature was kept below 50° by occasional cooling. When addition was completed, the reaction mixture was allowed to stand at room temperature for 45 min. The mixture was then added to a crushed ice-hydrochloric acid mixture, and the resulting precipitate of keto acid 2 (R = OCH<sub>3</sub>) collected by filtration. The crude product was dissolved in potassium carbonate solution, filtered and washed once with ether. The precipitate obtained upon acidification with concentrated hydrochloric acid was filtered, washed with water and dried to give 17.1 g (82%) of pure 2 (R = OCH<sub>3</sub>) as a white solid, mp 146°.

### ACKNOWLEDGEMENT

The authors take pleasure in acknowledging the experimental assistance of Mrs. M. Seydewitz and Messrs. R. Conrad, J. R. Conrad, R. N. Hellman, M. Horvath, R. F. Jodlowski, R. Kelly, R. G. Krawiec, R. Madel, R. Mehler, V. M. Poole Jr. and T. G. Soukup.

### REFERENCES

1. To whom correspondence should be addressed.
2. For a discussion of the succinoylation reaction, see E. Berliner in "Organic Reactions," Vol. V, John Wiley & Sons, Inc., New York, N. Y., 1949, Chapter 5 and A. G. Peto in Friedel-Crafts and Related Reactions," Vol. III, Part 1, G. A. Olah, Ed., Interscience Publishers, New York, N. Y., 1964, Chapter XXXIV.
3. (a) G. S. Krishna Rao and Sukh Dev, J. Ind. Chem. Soc., 34, 255 (1957).  
(b) W. G. Dauben and R. E. Adams, J. Am. Chem. Soc., 70, 1759 (1948).

3-(p-METHOXYBENZOYL)PROPIONIC ACID

- (c) K. W. Rosenmund and D. Shapiro, *Arch. Pharm.*, 272, 313 (1934); *Chem. Abstr.*, 28, 4046 (1934). (d) R. D. Haworth and G. Sheldrick, *J. Chem. Soc.*, 1950 (1934).
4. D. G. Thomas and A. H. Nathan, *J. Am. Chem. Soc.*, 70, 331 (1948).
  5. P. C. Mitter and S. De, *J. Ind. Chem. Soc.*, 16, 35 (1939).
  6. (a) L. F. Fieser, E. Berliner, F. J. Bondhus, F. C. Chang, W. G. Dauben, M. G. Ettliger, G. Fawaz, M. Fields, C. Heidelberger, H. Heymann, W. R. Vaughan, A. G. Wilson, E. Wilson, Mao-I Wu, M. T. Leffler, K. E. Hamlin, E. J. Matson, E. E. Moore, M. B. Moore, H. E. Zaugg, *J. Am. Chem. Soc.*, 70, 3197 (1948). (b) L. F. Fieser and V. Desreux, *ibid.*, 60, 2255 (1938). (c) L. F. Fieser and E. B. Hershberg, *ibid.*, 58, 2314 (1936). (d) C. C. Price and W. Kaplan, *ibid.*, 66, 477 (1944). (e) H. Plimmer, W. F. Short and P. Hill, *J. Chem. Soc.*, 694 (1938).
  7. O. Poppenberg, *Ber.*, 34, 3257 (1901).
  8. (a) G. Bargellini and M. Giua, *Gazz. Chim. Ital.*, 42, 197 (1912). (b) D. A. Hahn, *J. Am. Chem. Soc.*, 38, 1517 (1916).
  9. F. Krollpfeiffer and W. Schäfer, *Ber.*, 56, 620 (1923).
  10. A. S. Bailey, K. C. Bryant, R. A. Hancock, S. H. Morrell and J. C. Smith, *J. Inst. Petrol.*, 33, 503 (1947).
  11. S. Skaup and E. Schwamberger, *Ann.*, 462, 135 (1928).
  12. M. Tomita, J. Fushimi and M. Shibakawa, *J. Pharm. Soc. Jap.*, 64, 65 (1944); *Chem. Abstr.*, 45, 5660 (1951).

(Received April 27, 1971)